Fantastic properties of Microbubbles

AIST Dr.Masayoshi Takahashi

1

Contents of the presentation

- Fundamental properties of microbubble
- Gas hydrate formation
- Water remediation
- Waste water treatment

---- What is microbubble ? ----



How to make microbubbles (Examples)



Hydrodynamic type (High density type)





distilled water

Dissolved gas type



Bubble-size distribution (High density type)



10

Rising speed of microbubbles





Fundamental properties of microbubble for practical application

1. Increase in interior gas pressure



2. Increase in ion concentration around the gas-water interface



1. Increase in interior gas pressure



Young-Laplace equation

 $\Delta \mathbf{P} = 2\sigma/\mathbf{r}$

Increase in interior gas pressure



Gas hydrate formation

Gas hydrate = Ice-like material (with gas molecule) --- Low temperature / High pressure ---



Normal bubble (Conventional)

Microbubble 16



Temperature = 3 °C / 1 Atm.



Temperature = 1.1 °C / 1 Atm.



Pressure – Temperature diagram of Xe hydrate 19



Temperature

Remediation of water environmental by microbubble





Shio-Ashiya Port (Osaka bay area, Japan)

Microbubble generator for this test



Power :1.5 kW Water flow : 400L/min. Air supply : 10L/min.

The microbubble generator under sea



about 4m from the surface (1m from the bottom)

Photograph of the test site

Microbubble generator (about 4m from the surface)

and a second second second

The stream of microbubbles looks like white smoke

Before and after the microbubble treatment



Before / The world of death (with flash light)

At the bottom of the sea (-5m)



After about 3 months (without flash light)



Fundamental properties of microbubble for practical application

1. Increase in interior gas pressure



2. Increase in ion concentration around the gas-water interface

- Generation of free-radicals
- Generation of Nano-bubble



Electrical property of microbubble

 ζ potential



Smoluchowski's equation

$$\xi = \eta \ \mu \swarrow \varepsilon$$

$$\mu : \text{ the mobility } (\text{m}^2\text{s}^{-1}\text{V}^{-1})$$

$$\varepsilon : \text{ the dielectric constant } (\text{J}\text{V}^{-2}\text{cm}^{-1})$$

$$\eta : \text{ the viscosity of water } (\text{g}\text{cm}^{-1}\text{s}^{-1})$$

28

Movement of microbubbles in electrophoresis cell



<u>ζ potential of microbubble in distilled water</u>



The relationship between ζ potential and pH of the water



(adjustment by NaOH and NaCl)

Mechanism of bubble electricity



Change in ξ potential of microbubble during collapsing process



Free-radical generation by collapsing microbubble



Accumulated ions

Fundamental properties of microbubble for practical application

1. Increase in interior gas pressure

2. Increase in ion concentration around the gas-water interface



Generation of Nano-bubble

Measurement of free-radicals by ESR



Free-radical generation during collapsing process of microbubble



Experimental setup of radical generation





Experimental setup of radical generation



ESR spectrum



41

Experimental setup of phenol degradation



Degradation of phenol by collapsing air microbubble



The results of ESR test of ozone microbubble





Experimental setup of PVA degradation



Degradation of PVA by ozone microbubble



Practical application of collapsing microbubble

waste water treatment



Food industry (fishery product) : (200~300t/day)





To treat waste water from chemical factory

A phenol factory



Content	Waste water
Phenol	0.32 %
Formalin	0.56 %
Methanol	1.90 %
Acetone	0.08 %
n-butanol	0.03 %
nonvolatile	1.70 %

Not easy to treat by conventional methods

Collapse of ozone microbubble

Treatment of waste water from phenol factory





49

Development of new system to collapse microbubbles



Introduction of new collapsing system



Result of the test of microbubble treatment



Summary

Microbubble

- **—** Increase in the interior gas pressure
- Increase in the ion concentration around the gas-water interface
 - → Gas hydrate formation
 - → Environmental remediation
- **Collapse of microbubble**
 - -- Free-radical generation
 - -- Generation of Nano-bubble
 - → Waist water treatment

References

1) Takahashi, M. et al. Effect of shrinking microbubble on gas hydrate formation. J. Phys. Chem. B 107, 2171-2173(2003) 2) Takahashi, M. ζ potential of microbubbles in aqueous solutions: electrical properties of the gas-water interface. J. Phys. Chem. B 109, 21858-21864(2005) 3) Takahashi, M. Chiba, K. and Li, P. Free-radical generation from collapsing microbubbles in the absence of a dynamic stimulus. J. Phys. Chem. B 111, 1343-1347(2007) 4) Takahashi, M. Chiba, K. and Li, P. Formation of Hydroxyl Radicals by Collapsing Ozone Microbubbles under Strong Acid Conditions. J. Phys. Chem. B 111, 11443-11446(2007)